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### **Modeling of CO<sub>2</sub> Gas Detector Using Comsol Multiphysics**

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#### **ABSTRACT**

In these papers we have simulated carbon dioxide sensor using COMSOL Multiphysics (MEMS module). These sensors are cheap, highly sensitive and selective. We got the results of CO<sub>2</sub> detector electric potential variation at 300C temperatures with high dependability using less power.

**KEYWORDS:** COMSOL Multiphysics, CO<sub>2</sub> gas detector.

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## **INTRODUCTION**

The CO<sub>2</sub> gas in the atmosphere is one of the main reason for global warming, consequently monitoring CO<sub>2</sub> gas in the atmosphere is essential for good atmospheric conditions. To detect atmospheric carbon dioxide (CO<sub>2</sub>), the CO<sub>2</sub> sensor is required. Currently, in present world there are many CO<sub>2</sub> sensors are available, usually, they are fabricated by using the infrared sensing devices (IRD), but the magnitude of these sensors are relatively large and they required a very high power and temperature for the operation<sup>1</sup>.

In this paper, we suggested a designed CO<sub>2</sub> sensor in micrometer size (110 μm), which is designed by using a MEMS module of COMSOL Multiphysics software. The different type of CO<sub>2</sub> sensors presently used is Mixed Oxide Sensors<sup>2</sup>, NASICON-based sensors<sup>3</sup>, CNT sensors<sup>4</sup>, Gas Chromatographs and Spectroscopic Sensors (Non-Dispersive Infrared-NIRD). Similarly, in<sup>4</sup>, the aspect on wireless, passive Carbon Gas Sensor is given.

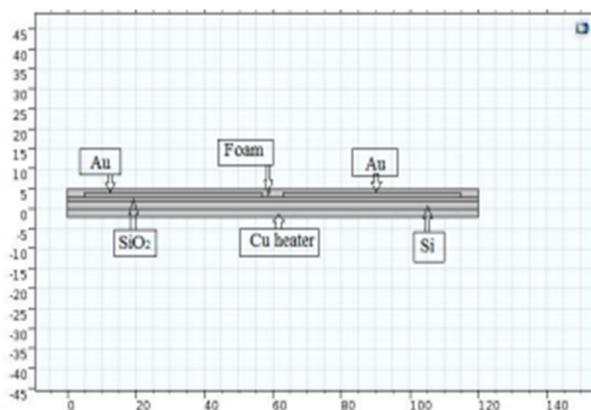
This type of detector has the disadvantage like high power consumption, would like high operational temperature, not easy for mobility circuits and sophisticated procedure. NIRD sensors needed additional power input in the order of mill watts and may price between fifty to five hundred USD. Potentiometric mixed chemical compound detectors and NASICON based sensor are quite tough to induce stabilized activity<sup>5</sup>, this variety of sensors measure to rely on concentration of gas and operational temperature ought to be quite 350 C. The greenhouse gas detector is often wont to calibrate warmth performance of the sensor; numerous sensors forever have thermal inertia and finite heat electrical phenomenon, which alter the waveform of the Temperature versus Time<sup>6</sup>. At present, it is essential to provide high-quality CO<sub>2</sub> gas sensors for many fields like food industries, agriculture industries, air-conditioning, and atmospheric monitoring due to the massive rise of CO<sub>2</sub> levels in the atmospheric because of greenhouse effects<sup>7</sup>. Solid state potentiometric CO<sub>2</sub> sensors are precisely favorable to detect low levels of CO<sub>2</sub> concentrations; also they provide the firm reaction, ease, and long term stability. The sensor's electromotive force (emf) displays a linear relationship with the logarithm of CO<sub>2</sub> concentration based on solid electrolyte for an air quality control system<sup>8</sup>. In Electric noses for observing environmental pollution and building relapse model arranged the detail case study about the present system for examining the use of different sensors grid system concerning urban air fumes monitoring for carbon monoxide, carbon dioxide (CO, CO<sub>2</sub>) gases for three different regions in Alexandria- Egypt and 2 different traffic roads<sup>9</sup>. In solid-state potentiometric CO<sub>2</sub> sensor in dense film technology for breath examination described with the detail observing of the CO<sub>2</sub> gas, the concentration which is valuable in numerous applications, for example, to define the air quality, to monitor food quality or for breath analysis. In it is projected a low-power-consumption CO<sub>2</sub> gas sensor using ionic liquids for green energy management. They study of the

impedance of the ionic liquid material having low concentration of the CO<sub>2</sub> gas which increases in the air. In this paper, we suggested a sensor which protects the Gold electric terminals of the sensor by dropping the foam drop over the sensor electrodes, which reduce the atmospheric oxygen effect on the sensor performance. In this sensor model,<sup>10</sup> we examine the electric potential variation with respect to the arc length plotted on the top surface gold electrodes of the CO<sub>2</sub> sensor. The sensor resides of an interdigitated transducer (IDT) etched onto a piezoelectric SiO<sub>2</sub> substrate and enclosed with the foam material. The mass of the foam material increases as it increases the absorption of CO<sub>2</sub> from the air. The bottom layer of copper material is exertion as a heater for CO<sub>2</sub> sensor<sup>11</sup>.

## MODEL DEFINITION

### *The Geometry of the CO<sub>2</sub> Detector*

Fig. 1 shows a cross-sectional view of the CO<sub>2</sub> gas detector in the COMSOL Multiphysics simulation software. The gold electrodes are fixed on both cross surfaces of SiO<sub>2</sub> and Si-surface film. The higher exterior is painted by a BF<sub>4</sub> foam material to guard the electro-active portion of the device<sup>12</sup>, the globule then covers the whole surface of the sensor. BF<sub>4</sub> is the best.



**Figure I: Geometry and Dimensions of the CO<sub>2</sub> detector**

Fig.1. also, demonstrations the dimensional view of the detector, the height of the sensor is 7μm and the length of the sensor is 120μm. The gap between the two electrodes is 10μm and from both the border side it is 5 μm distances maintained.

### *Boundary Conditions*

The model is simulated underneath with flexible material over the piezoelectric material and the foam is at the higher layer of the model<sup>13</sup>. Foam is used to captivate the CO<sub>2</sub> gas. The model has an intervallic boundary condition at the vertical edge and simulated for Eigen frequency as a solver<sup>14</sup>. Solid mechanics, electric current, electrostatics, and heat transfer physics are sequentially solved in the geometrical conduction.

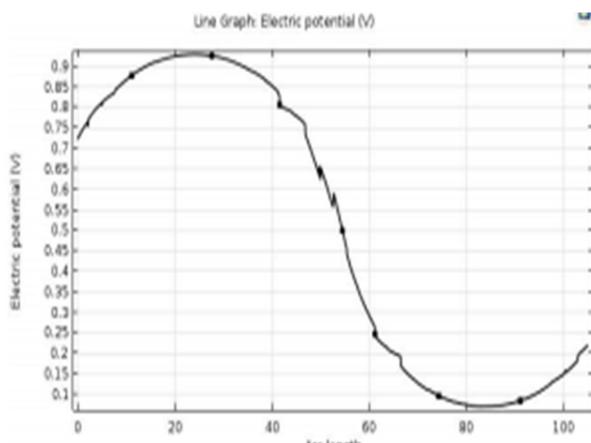


Figure II: Electric Potential VS Field

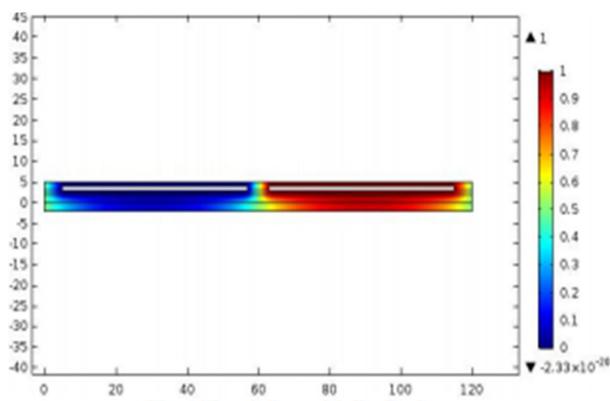


Figure III: Thermal image

## RESULT AND DISCUSSION

The irreversible reaction of  $\text{SiO}_2$  and  $\text{CO}_2$  is given in (4). Where, produced  $\text{CO}_2$  is negatively charged. When  $\text{CO}_2$  get adsorb on  $\text{SiO}_2$  layer electrons are transferred from  $\text{SiO}_2$  to  $\text{CO}_2$  i.e.  $\text{CO}_2$  extract electrons from  $\text{SiO}_2$ . As electrons are transferred from  $\text{SiO}_2$ , the potential of  $\text{SiO}_2$  is increases (\*). Potential distribution in  $\text{SiO}_2$  is increases with increase in concentration of  $\text{CO}_2$ . The basic cause of impedance varies with  $\text{CO}_2$  concentration is shown by following graph decreases the overall cost of the sensor. The optimum design is simulated and its deformation effect.

## CONCLUSION

Nowadays, considerable amount of work is being carried out in order to develop sensors that are more cost efficient, simple to use, sensitive and selective. Research efforts have developed gas sensors for inflammable and toxic gases. Sensors are all around us and their prevalence is growing rapidly. Sensors play a very important role in the modern society with all the modern technology around us.  $\text{CO}_2$  sensor is cheap and can be easily fabricated.

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